A General Method for the Preparation of Cis-Annelated Rings

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(Received in UK 4 December 1973; accepted for publication 18 December 1973) The ability of certain bridged bicyclic olefins to undergo the Cope rearrangement¹ or a formal oxy- Cope rearrangement² suggested a general approach to 5/6, 6/6 and 5/7 cis-fused ring systems which occur in numerous naturally occurring terpenoids.³ This approach involves the Cope rearrangement (1+2) of the vinylbicycloolefins (1; X=CH₂) which provides <u>cis</u>-bicyclic products (2) with unsaturated functionality capable of further useful modification.

The vinyl bicycloolefins (1; $X=CH_2$; m=1, n=1 or 2) were readily available by Wittig reaction from the corresponding ketones (1; X=O) which were, in turn, prepared by Diels-Alder reactions of cyclopentadiene and cyclohexadienes with vinyl ketones. Diels-Alder reactions involving cyclopentadiene gave predominately (>80%) the <u>endo</u>-adducts (1; X=O, m=n=1, R=Me, i-Pr, t-Bu and Ph) which, on base catalysed equilibration (NaOMe/MeOH) gave values of K _{exo/endo} ranging from 0.98 (1; X=O, m=n=1, R=Me)[†] to 2.96 (1; X=O, m=n=1, R=t-Bu). The steric requirements of the system (1; X=O, m=n=1) resulted in an inefficient Wittig reaction, in most cases, when ether or tetrahydrofuran was used as solvent. Wittig reactions in DMSO gave better yields but resulted in exo-endo equilibration.^{††}

+This value differs from that previously reported.⁴

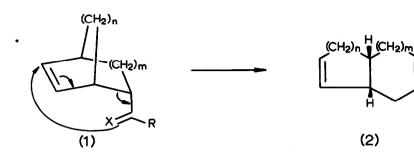
⁺⁺Such epimerisation problems need not present insurmountable synthetic problems and may be overcome by replacing the epimerisable hydrogen by a suitable substituent.

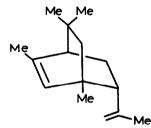
Similar results were obtained in the bicyclo [2.2.2] octene series (1; X=O or CH₂, n=2, m=1) but in this case the endo isomers predominated on equilibration.

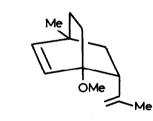
The <u>endo</u>-isomers (1; $X=CH_2$, m=n=1, R=Me, Et, i-Pr or Ph) were all cleanly and quantitatively isomerised at ca. 150[°] to the corresponding 5/6 systems (2; m=n=1, R=Me, Et, i-Pr or Ph). In all cases the corresponding <u>exo</u>-alkenyl compounds remained unchanged under the same conditions.

Similar results were obtained with the bicyclo [2.2.2] octenes (1; X=CH₂, m=1, n=2, R=Me), (3) and (4) which rearranged to the corresponding 6/6 systems (2; m=1, n=2, R=Me), (5; R=R¹=R²=Me, R³=H) and (5; R=R²=H, R¹=OMe, R³=Me). The enol ether (5; R=R²=H, R¹=OMe, R³=Me) was readily hydrolysed to the corresponding ketone (6) in the course of chromatography on silica. Temperatures required to effect the Cope rearrangement in the bicyclo [2.2.2.] octene series were typically 220-255° and although the <u>exo</u>-isomers of (3) and (4) were unchanged under the conditions required to rearrange their <u>endo</u>-isomers, the <u>exo</u>-isomer of (1; X=CH₂, m=1, n=2, R=Me) was slowly consumed at 225°⁵. N.m.r. spectra were particularly useful in monitoring the reactions, e.g. the characteristic olefinic proton pattern of (4; τ , 5.02, q, 2H,J_{AB}=9Hz, and terminal methylene protons as a broad singlet at τ 5.29) was replaced by that of (5; R=R²=H, R¹=OMe, R³=Me; broad singlet (1H) at τ 4.78 and vinyl ether proton as a broad triplet at τ 5.67).

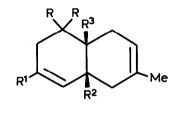
Our approach to the synthesis of the 5/7 bicyclic system takes advantage of the ready availability of <u>endo-2-acylbicyclo</u> [2.2.2] oct-5-enes previously utilised for the construction of the 6/6 bicyclic systems. Photo-chemical 2+2 cycloaddition (Paterno-Buchi reaction)^{6,7} of (7; a-c) gave the corresponding







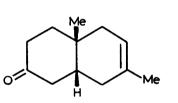
(4)



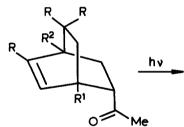
(5)

R²

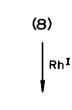
(3)







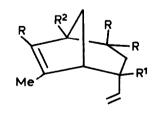
(7) (a) $R = R^1 = R^2 = H$ (b) $R = R^1 = Me, R^2 = H$ (c) $R = H, R^1 = OMe, R^2 = Me$

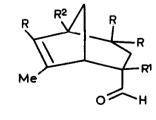


O

R

Me





(10)

(9)

R

oxetanes (8; a-c) in high yield. Cleavage of these oxetanes with $Rh_2(CO)_4Cl_2$ gave the endo-2-formylbicyclo [3.2.1] oct-6-enes (9; a-c).⁷ Two of these (9; a and b) have, thus far, been converted by Wittig reaction to the corresponding vinyl derivatives (10; a and b). Thermal rearrangement of (10a) occurred at 232° (14 h., neat liquid) to give one principal product and a considerable amount of unchanged starting material (55:45). G.l.c. analysis also revealed ca. 10% of minor products. A similar result was obtained at 260° in benzene solution. In contrast (10b) was unchanged at 300°. The product from (10a) is currently under investigation.

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